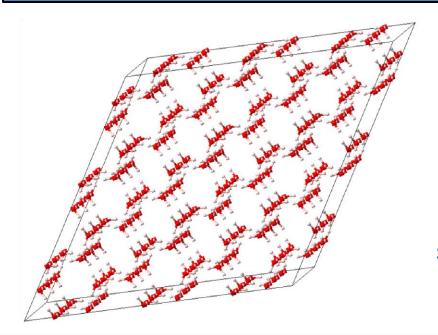
Benchmarking a First-Principles Thermal Neutron Scattering Law for Water Ice with a Diffusion Experiment



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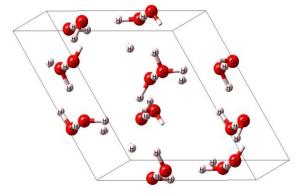
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Motivation (why ice?)

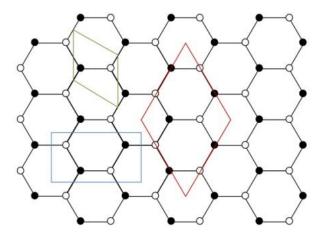
- 1. Criticality safety analyses routinely examine the reactivity effect of a liquid water environment during the transport and storage of nuclear fuel material.
- 2. For cold weather conditions, it is of interest to have a thermal neutron scattering law (TSL) for water ice which characterizes the impact of the chemically bound crystalline structure on neutron scattering and reflection.
- 3. The thermal scattering kernel for liquid water has been extensively studied and is available in an ENDF library. However, no published ENDF File 7 TSL for water ice exists.
- 4. Crystalline ice is a much better reflector than liquid water, although it is a poorer moderator than liquid water. Modifying the material density of liquid water to account for ice formation will not capture these effects in benchmark calculations.

The Structure of Hexagonal Ice Ih

- H₂O molecules are bound by strong covalent bonds internally. Weak intermolecular hydrogen bonds allow for many rotational configurations of the H₂O molecules in ice on a fixed oxygen lattice.
- Ice Ih is the most common form of H₂O ice. This structure can be viewed as two superimposed hexagonal lattices, where each lattice site contains an oxygen. The molecular alignments are locally ordered but globally disordered.
- Ice Ih has 4 molecules per unit cell (green trapezoid). Larger unit cells can be defined to better capture the varying H₂O alignments. The red diamond is a 2-D view of the 12-molecule 3-D unit cell used in this work (top right image).



3-D view of the ice Ih structure.



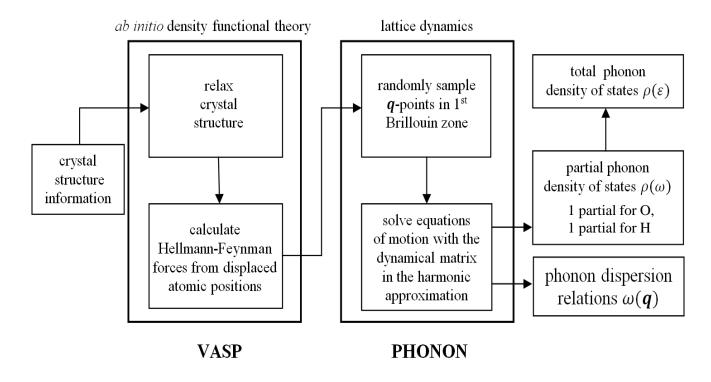
2-D view of the ice I*h* oxygen lattice.

Objectives

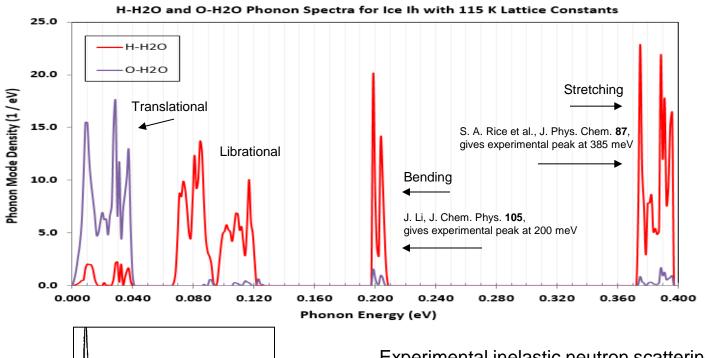
- 1. Use the VASP *ab initio* density functional theory (DFT) code and the PHONON lattice dynamics code to model the interatomic structure of ice Ih and its dynamical behavior.
- 2. Calculate the vibrational phonon energy spectrum for ice Ih and use this to calculate the $S(\alpha,\beta)$ TSL and integral inelastic scattering cross sections in the incoherent approximation.
- 3. Add in absorption and integral incoherent elastic scattering cross sections to determine the total cross sections as a function of temperature.
- 4. Based on experimentally measured inelastic neutron scattering spectra and total cross sections, optimize the phonon spectrum to account for the globally averaged ice Ih structure.
- 5. Benchmark the ice Ih thermal scattering kernel using MC21 against measured flux decay time eigenvalues from a pulsed-neutron die-away neutron diffusion experiment for cylinders of ice with various geometries over a range of temperatures.

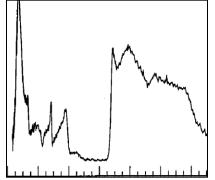
Determining the Phonon Vibrational Energy Spectrum

- Ab initio method of modeling the structure and vibrational energy spectrum of ice Ih using density functional theory (DFT) and lattice dynamics.
- High-accuracy experimental lattice constants for Ice Ih were enforced to correctly model the hydrogen bond length.



Calculated Phonon Energy Spectra





Experimental inelastic neutron scattering spectrum for polycrystalline ice Ih (same energy scale as above).

J. Li, "Inelastic Neutron Scattering Studies of Hydrogen Bonding in Ices," *Journal of Chemical Physics* **105**, 6733-6755 (1996).

Calculating the Thermal Neutron Scattering Kernel for Ice Ih

$$\frac{d^2\sigma(E)}{d\alpha d\beta} = \sigma_{\rm b} F[S(\alpha, \beta)]$$

double-differential cross section (in angle and in energy)

 α

B

momentum transfer factor (scattering angle dependence)

energy transfer factor

$$\frac{d\sigma(E)}{dE'} = \frac{1}{k_{\rm B}T} \frac{d\sigma(E)}{d\beta} = \int_{\alpha_{\rm min}(E,\beta,T)}^{\alpha_{\rm max}(E,\beta,T)} \frac{d^2\sigma(E)}{d\alpha d\beta} d\alpha$$

differential cross section in energy

$$\sigma(E) = \int_{\beta_{\min}(E,T)}^{\infty} \int_{\alpha_{\min}(E,\beta,T)}^{\alpha_{\max}(E,\beta,T)} \frac{d^2 \sigma(E)}{d\alpha d\beta} d\alpha d\beta$$

integral inelastic cross section

The phonon energy spectrum is the fundamental parameter determining the thermal scattering law $S(\alpha,\beta)$ and incoherent elastic scattering for Ice Ih.

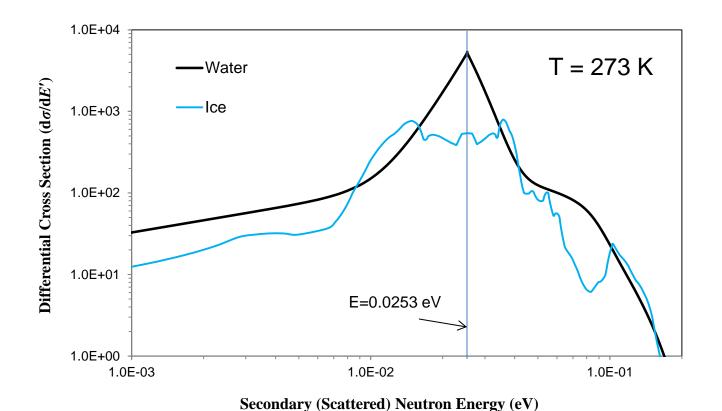
$$\sigma(E) = \frac{\sigma_{\text{incoh}}}{2} \left(\frac{1 - e^{-4WE}}{2WE} \right)$$

integral incoherent elastic cross section

All cross section calculations assume the incoherent approximation. Since the nuclear bound scattering cross section for H_2O is very strongly incoherent, this is expected to yield good results for ice Ih.

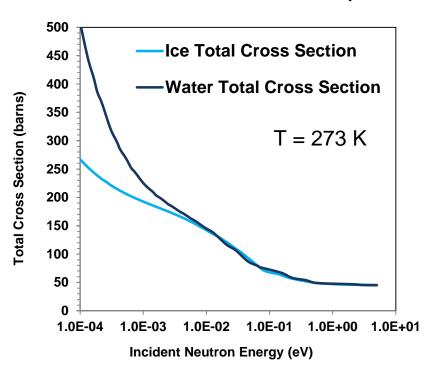
Total cross section = inelastic + elastic + absorption

Comparison of the Secondary Neutron Energy Distribution for Inelastic Scattering with Hydrogen in Water and Ice

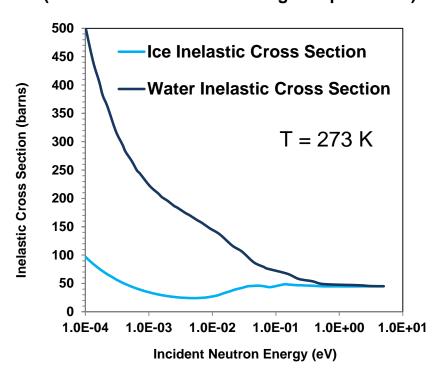


Comparison of Total and Inelastic Scattering Cross Sections for Water and Ice

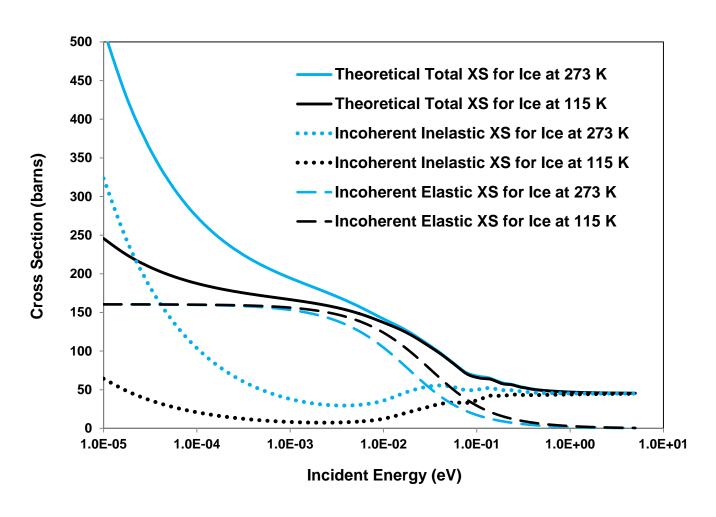
Total = Inelastic + Elastic + Absorption



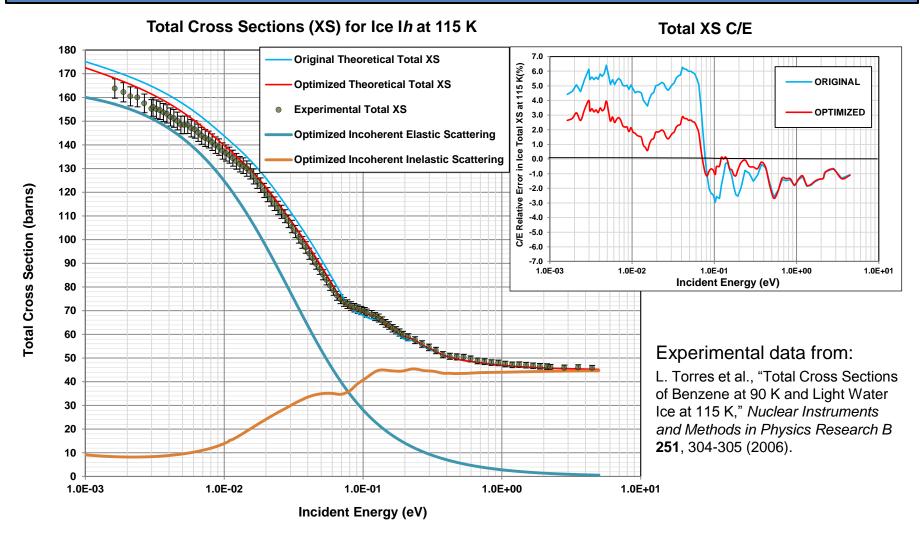
Inelastic Only (There is no elastic scattering in liquid water)



Components of Thermal Neutron Scattering Cross Sections for Ice Ih at 115 K and 273 K



Comparison of Theoretical and Experimental Total Cross Sections for Ice Ih



Benchmarking the Ice Ih Thermal Scattering Kernel with a Neutron Diffusion Experiment

- It is desirable to test the performance of the thermal neutron scattering kernel for Ice Ih in an integral benchmark over a range of temperatures. As no critical benchmarks designed to test water ice exist, a neutron diffusion benchmark will be incorporated.
- E. G. Silver measured the diffusion parameters of Ice Ih in the 1960s using a pulsed-neutron dieaway experiment to determine the fundamental mode flux decay time eigenvalues for various-sized cylinders of ice over a range of temperatures.
- MC21 is used to test the Ice Ih scattering kernel. NDEX reads ENDF File 7 and generates a thermal library for MC21. The ice cylinders were modeled and the resulting time eigenvalues for flux decay after an initial neutron pulse were determined and compared to experiment.
- Finally, to test the sensitivity of the benchmark to a correct scattering kernel for Ice Ih, additional tests were performed using a subcooled liquid water library at ice density and using a free gas library at ice density.

Benchmark experimental reference:

E. G. Silver, "A Pulsed-Neutron Investigation of the Effect of Temperature on the Decay of a Thermal-Neutron Population in H₂O Ice," *Nuclear Science and Engineering* **34**, 275-284 (1968).

The Diffusion Experiment and Diffusion Theory

- A high-voltage deuteron accelerator was used with a deuterium target placed adjacent to the ice cylinder. A pulse of high-energy neutrons was produced and an adjacent lithium iodide detector and time channel analyzer were used to record the flux decay. Data was collected over many pulses.
- After sufficient waiting time following the initial pulse, the neutron population is in thermal equilibrium and settles into the fundamental spatial mode with a single flux decay time eigenvalue.
- In the one-speed diffusion model, the flux decay of the fundamental spatial mode can be expressed as

 $\varphi(\mathbf{r},t) = \varphi_0(\mathbf{r}) \exp[-(v\Sigma_a + vDB^2 - CB^4)t]$, where the time eigenvalue $\alpha = v\Sigma_a + vDB^2 - CB^4$

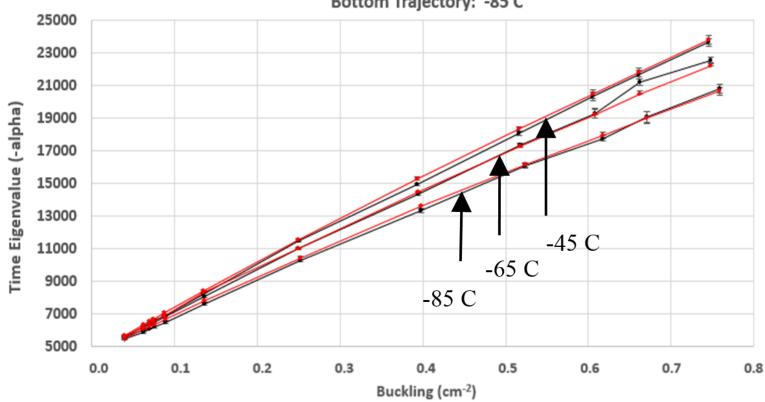
 Σ_a is the macroscopic absorption cross section D = $1/\Sigma_{transport}$ is the diffusion parameter, v = the effective average neutron velocity, B² = geometric buckling, and C = the diffusion cooling parameter

While the analytical form for the α time eigenvalue is approximate, we should expect that small geometries with high buckling should be very sensitive to the TSL supplied and large geometries with low buckling should be relatively insensitive to the TSL.

Benchmark Testing Results

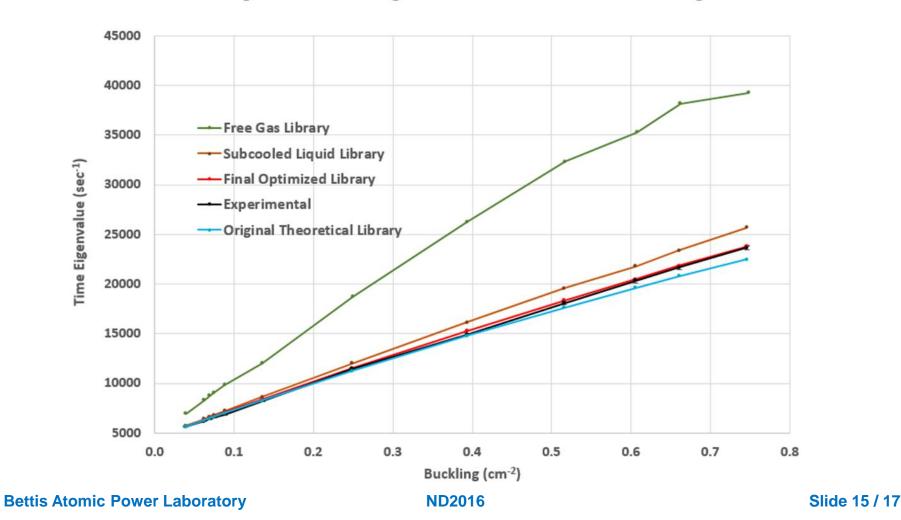
Experimental (Black) vs. Theoretical (Red) Eigenvalues vs. Buckling

Top Trajectory: -45 C Middle Trajectory: -65 C Bottom Trajectory: -85 C



Benchmark Testing Results

Time Eigenvalues vs. Buckling at -45 C for Selected Theoretical Scattering Kernels



Benchmark Testing Results

| SILVER EXPERIMENTAL | | | | | | | | | FINAL LIBRARY (-45 C) | | | | FINAL LIBRARY (-65 C) | | | | FINAL LIBRARY (-85 C) | | | |
|---------------------|-------|-----|----------------|-------|-----|----------------|-------|--------|-----------------------------|---------|--------|--------|-----------------------------|---------|--------|--------|-----------------------------|---------|--------|-------|
| -45 C | | | -65 C | | | -85 C | | MC21 α | σ | C/E | σ | MC21 α | σ | C/E | σ | MC21 α | σ | C/E | σ | |
| B ² | α | σ | B ² | α | σ | B ² | α | σ | | | | | | | | | | | | |
| 0.0394 | 5630 | 52 | 0.0395 | 5579 | 59 | 0.0396 | 5443 | 35 | 5659 | 6 | 1.0051 | 0.009 | 5589 | 3 | 1.0018 | 0.009 | 5504 | 3 | 1.0112 | 0.007 |
| 0.0621 | 6207 | 38 | 0.0621 | 6031 | 35 | 0.0624 | 5890 | 61 | 6325 | 5 | 1.0191 | 0.006 | 6190 | 4 | 1.0264 | 0.006 | 6057 | 1 | 1.0284 | 0.011 |
| 0.0689 | 6413 | 66 | 0.0690 | 6266 | 22 | 0.0693 | 6096 | 53 | 6523 | 2 | 1.0172 | 0.010 | 6370 | 2 | 1.0166 | 0.010 | 6217 | 4 | 1.0199 | 0.009 |
| 0.0748 | 6584 | 70 | 0.0749 | 6455 | 38 | 0.0752 | 6226 | 66 | 6697 | 2 | 1.0171 | 0.011 | 6530 | 2 | 1.0117 | 0.011 | 6365 | 2 | 1.0223 | 0.011 |
| 0.0880 | 6888 | 49 | 0.0881 | 6825 | 76 | 0.0885 | 6493 | 72 | 7074 | 3 | 1.0270 | 0.007 | 6876 | 3 | 1.0075 | 0.007 | 6682 | 2 | 1.0290 | 0.011 |
| 0.1352 | 8214 | 70 | 0.1354 | 8052 | 73 | 0.1362 | 7593 | 72 | 8415 | 3 | 1.0244 | 0.009 | 8356 | 5 | 1.0377 | 0.009 | 7804 | 3 | 1.0278 | 0.010 |
| | | | | | | | | | Low B ² Avg. C/E | | 1.0183 | | Low B ² Avg. C/E | | 1.0169 | | Low B ² Avg. C/E | | 1.0231 | |
| 0.2490 | 11493 | 87 | 0.2494 | 10997 | 76 | 0.2514 | 10272 | 82 | 11546 | 12 | 1.0046 | 0.008 | 10985 | 10 | 0.9989 | 0.008 | 10427 | 8 | 1.0150 | 0.008 |
| 0.3933 | 14915 | 73 | 0.3941 | 14341 | 65 | 0.3981 | 13310 | 100 | 15302 | 56 | 1.0259 | 0.006 | 14458 | 43 | 1.0082 | 0.006 | 13598 | 33 | 1.0216 | 0.008 |
| 0.5161 | 18083 | 127 | 0.5174 | 17307 | 132 | 0.5233 | 16055 | 129 | 18350 | 59 | 1.0148 | 0.008 | 17259 | 45 | 0.9972 | 0.008 | 16158 | 47 | 1.0064 | 0.009 |
| 0.6058 | 20349 | 245 | 0.6078 | 19292 | 295 | 0.6172 | 17772 | 189 | 20482 | 52 | 1.0065 | 0.012 | 19232 | 39 | 0.9969 | 0.012 | 17952 | 30 | 1.0102 | 0.011 |
| 0.6606 | 21669 | 258 | 0.6624 | 21226 | 195 | 0.6710 | 19073 | 344 | 21818 | 104 | 1.0069 | 0.013 | 20449 | 77 | 0.9634 | 0.013 | 19033 | 57 | 0.9979 | 0.018 |
| 0.7460 | 23662 | 229 | 0.7482 | 22555 | 198 | 0.7586 | 20808 | 236 | 23795 | 105 | 1.0056 | 0.011 | 22199 | 77 | 0.9842 | 0.011 | 20663 | 57 | 0.9930 | 0.012 |
| | | | | | | | | | High B ² A | wg. C/E | 1.0107 | | High B ² A | vg. C/E | 0.9915 | | High B ² A | vg. C/E | 1.0074 | |

For the six highest buckling geometries (B² is in units of cm⁻²), which are the most sensitive to the supplied thermal scattering kernel, the average C/E across all three temperatures is 1.003.

For the six lowest buckling geometries, the average C/E across all three temperatures is 1.019. As these cases are relatively insensitive to the supplied thermal scattering kernel, this indicates there may be a slight positive bias in the absorption cross sections. Indeed, this is consistent with the ~2.5% ENDF quoted uncertainty for H absorption cross sections in the thermal energy range.

Summary / Conclusions

- While the total cross sections for liquid water and solid ice are similar (except at
 extremely low incident energies), the details of the scattering kernels differ significantly
 and result in significant differences in the diffusion of neutrons through the material.
- Unlike liquid water, a large fraction of neutron scattering that takes place in ice is elastic.
 Consequently, Ice Ih has enhanced reflective properties over liquid water that could have important criticality safety implications.
- Combining the first-principles VASP/PHONON methodology with experimentally measured scattering data and a diffusion benchmark has resulted in the development and successful testing of a thermal neutron scattering kernel for Ice Ih.
- A full ENDF File 7 thermal neutron scattering library will be submitted to the National Nuclear Data Center for the ice Ih structure for a range of temperatures of interest.